Following the normal protocol, this should imitate the results of investigating the effect of phosphates on phosphatase experiment well. Care should be taken to ensure that lids for sodium carbonate bottles are not put on any buffer bottles by accident, as this will trigger the phenolphthalein to turn pink.

Preparation of buffers for PPP experiment alternative are as follows:-

Phenolphthalein prepared as follows- Dissolve 0.5 g in 50 cm³ of ethanol (FLAMMABLE) and make up to 100 cm³ with water. Label the container as FLAMMABLE.

Phosphate Buffer (pH5) 0M- 20 drops of Phenolphthalein in 150ml water.

Phosphate Buffer (pH5) 0.05M- 10 drops of Phenolphthalein in 150ml water.

Phosphate Buffer (pH5) 0.10M- 5 drops of Phenolphthalein in 150ml water.

Phosphate Buffer (pH5) 0.20M- 2 drops of Phenolphthalein in 150ml water.

Phosphate Buffer (pH5) 0.30M-1 drop of Phenolphthalein in 150ml water.

If using negative control add distilled water with **<u>no</u>** phenolphthalein.

Sodium carbonate prepared as normal.

PPP prepared as water.

Phosphatase extraction carried out as normal to simulate proper experiment.

Photo of expected results:-



0M 0.05M 0.1M 0.2M 0.3M



Phosphatase Enzymes in Plants

Read these instructions carefully before you start.



In step 5, the following reaction occurs:



7. The addition of sodium carbonate solution stops the reaction and, if free phenolphthalein is present, a pink colour is seen.

8. Measure the pink colour in a colorimeter using a blank with 1cm3 water in place of the supernatant. If you have no colorimeter, you can compare your samples with a serial dilution of lab phenolphthalein, to give you an approximate estimate of the amount of phenolphthalein released by the enzyme.